# Isotopes Practice Set

1. What is an isotope? _____

2. What does the number next to isotopes mean? _____

3. How can you tell isotopes apart? _____

For each of the following isotopes, write the number of protons, neutrons, and electrons.

|  | Chromium-58 | Chromium-63 |
|---|---|---|
| # of protons |  |  |
| # of neutrons |  |  |
| # of electrons |  |  |

|  | Carbon-12 | Carbon-14 |
|---|---|---|
| # of protons |  |  |
| # of neutrons |  |  |
| # of electrons |  |  |

|  | Nitrogen-15 | Nitrogen-20 |
|---|---|---|
| # of protons |  |  |
| # of neutrons |  |  |
| # of electrons |  |  |

|  | Sulfur-33 | Sulfur-36 |
|---|---|---|
| # of protons |  |  |
| # of neutrons |  |  |
| # of electrons |  |  |

Fill in the isotope names and any missing information including isotope numbers from the chart. Use your periodic table and the information provided.

|  | Iodine-127 | Iodine-131 |
|---|---|---|
| # of protons |  |  |
| # of neutrons |  |  |
| # of electrons |  |  |

|  |  |  |
|---|---|---|
| # of protons |  |  |
| # of neutrons |  |  |
| # of electrons |  |  |

|  |  |  |
|---|---|---|
| # of protons |  |  |
| # of neutrons |  |  |
| # of electrons |  |  |

|  |  |  |
|---|---|---|
| # of protons |  |  |
| # of neutrons |  |  |
| # of electrons |  |  |

# [Isotope Practice Set Answers](#)

**L Darling-Hammond**

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