



Naming Acids Worksheet Packet

Name: _____ Answer Key

1. What is the definition of an acid?

An acid is a substance that when in water gives off hydrogen ions

2. What are the rules for naming acids?

If no oxygen in anion HYDRO + root of anion + IC acid

If oxygen add - IC to root of anion or element if -ate ending on anion or add -OUS to root of anion or element if -ite ending on anion.

3. Write the names and the anions for the following acids.

<u>Acid</u>	<u>Anion</u>	<u>Name</u>
H_2SO_4	sulfate	sulfuric acid
H_3PO_4	phosphate	phosphoric acid
$\text{HC}_2\text{H}_3\text{O}_2$	acetate	acetic acid
H_2SO_3	sulfite	sulfurous acid
HNO_2	nitrite	nitrous acid
HClO_4	perchlorate	perchloric acid
HClO_3	chlorate	chloric acid
HClO_2	chlorite	chlorous acid
HClO	hypochlorite	hypochlorous acid

Naming Acids Packet

Karin Nielsen-Saines



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