Naming and Equations Review

Name the following compounds. Remember, they may be either ionic or covalent compounds, so make sure you use the right naming method!

- 1) LiBr _____
- 2) Na₂O ______
- 3) CaSO₄_____
- 4) Be(NO₃)₂ _____
- 5) NO _____

Write the formulas for the following compounds. Remember, they may be either ionic or covalent compounds, so make sure you use the right method!

- 6) sodium nitride _____
- 7) sulfur dioxide _____
- 8) ammonia _____
- 9) calcium phosphate _____
- 10) aluminum nitrate _____

Balance the following equations:

- 11) ____ NaCN + ___ CuCO₃ → ___ Na₂CO₃ + ___ Cu(CN)₂
- 12) ____ Al(OH)₃ + ____ Cs → ___ CsOH + ___ Al
- 13) ____ MgS + ___ OF₂ \rightarrow ___ MgF₂ + ___ SO
- 14) $_{--}$ S₈- * $_{--}$ H₂ \rightarrow $_{--}$ H₂S
- 15) ____ LiCl + ____ Br₂ → ____ LiBr + ____ Cl₂
- 16) ____ $Pb(NO_3)_2 + ___ HF \rightarrow ___ HNO_3 + ___ PbF_2$
- 17) ____ GaBr₃ + ____ Na₂SO₄ \rightarrow ____ NaBr + ___ Ga₂(SO₄)₃
- 18) ____ P₂S₂ → ___ P₄ + ___ Sa

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Steven S. Zumdahl

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