

১৪। কোন ধরনের শিক্ষার মাধ্যমে মানবসম্পত্তি প্রতিরোধ সম্ভব?

- (ক) ধর্মীয় ও সামাজিক (খ) ধর্মীয় ও নৈতিক  
(গ) নৈতিক ও সামাজিক (ঘ) সামাজিক ও মানসিক

১৫। মানকে আসক্ত ব্যক্তির নিজের ক্ষতির পাশাপাশি কার ক্ষতি করে?

- (ক) আত্মীয় বন্ধনের (খ) নিকট আত্মীয়ের  
(গ) প্রতিবেশীর (ঘ) সমাজ ও রাষ্ট্রের

১৬। বাংলাদেশের জনসংখ্যা নীতির মূল লক্ষ্য ও উদ্দেশ্য কয়টি?

- (ক) ৫টি (খ) ৭টি  
(গ) ৯টি (ঘ) ১১টি

১৭। যে কারণে আগের তুলনায় বর্তমানে জনসংখ্যা বৃদ্ধির হার কমে এসেছে—

- (i) বিনেশি সহায়তা (ii) পরিবার পরিকল্পনা  
(iii) মানুষের ক্রমবর্ধমান সচেতনতা

নিচের কোনটি সঠিক?

- (ক) i ও ii (খ) ii ও iii  
(গ) i ও iii (ঘ) i, ii ও iii

১৮। মারমারা কোন ধর্মের অনুসারী?

- (ক) বৌদ্ধ (খ) হিন্দু  
(গ) খ্রিস্টান (ঘ) ইসলাম

১৯। সাংগাই উৎসব পালন করা হয় কখন?

- (ক) চৈত্র সংক্রান্তিতে (খ) পৌষ পার্বণে  
(গ) বিয়ের অনুষ্ঠানে (ঘ) নীলা লাভের অনুষ্ঠানে

২০। রাজবংশী কী?

- (ক) বৌদ্ধদের সনাতন নাম (খ) সূর্য দেবতার নাম  
(গ) আবাইনদের অগ্নি নাম (ঘ) ক্ষুদ্র জাতিসত্তার নাম

২১। UNESCO এর সদর দপ্তর কোথায়?

- (ক) জেনেভা (খ) প্যারিস  
(গ) নিউইয়র্ক (ঘ) জেন্না

২২। UNICEF যে লক্ষ্যে কাজ করে—

- (i) মাতৃ ও শিশু স্বাস্থ্য  
(ii) প্রাথমিক স্বাস্থ্য পরিচর্যা  
(iii) পুষ্টি ও রোগ প্রতিরোধ

নিচের কোনটি সঠিক?

- (ক) i ও ii (খ) ii ও iii  
(গ) i ও iii (ঘ) i, ii ও iii

২৩। নিচের কোনটি বনিজ সম্পদ?

- (ক) পানি (খ) সিমেন্ট  
(গ) টিনামাটি (ঘ) জৈবসার

২৪। বাংলাদেশে প্রথম চিনিকল কোনটি?

- (ক) রংপুর চিনিকল (খ) রাজশাহী চিনিকল  
(গ) চন্দ্রঘোনা চিনিকল (ঘ) নাটোর গোপালপুর চিনিকল

\* অনুচ্ছেদটি পড়ে ২৫ ও ২৬ নং প্রশ্নের উত্তর দাও :  
বাংলাদেশে উর্বর মাটি, নদ-নদী এবং নানা প্রজাতির প্রাণী রয়েছে। সুষ্ঠু পরিকল্পনা প্রণয়ন করে এগুলো ব্যবহার করতে পারলে এ দেশ ঠিকই সমৃদ্ধ হবে।

২৫। উদ্দীপকে কোন বিষয়টির ইঙ্গিত পাওয়া যায়?

- (ক) সাংস্কৃতিক নিক (খ) প্রাকৃতিক সম্পদ  
(গ) মানবসৃষ্ট সম্পদ (ঘ) সঠিক সম্পদ

২৬। এ দেশ সমৃদ্ধ হতে পারে যদি উদ্ভেদিত উপাদানগুলোর—

- (ক) সচেতন ব্যবহার নিশ্চিত হয়  
(খ) খুব কম ব্যবহার হয়  
(গ) শিকিত সমাজ হয়  
(ঘ) খুব ধীরে ধীরে ব্যবহার হয়

২৭। বর্তমান বিশ্ব কী নির্ভর?

- (ক) মোবাইল (খ) শিক্ষা  
(গ) শিল্প (ঘ) প্রযুক্তি

২৮। মুজিব নগর সরকারের রাষ্ট্রপতি ছিলেন—

- (ক) বলাবকু শেখ মুজিবুর রহমান  
(খ) তালউদ্দীন আহমদ  
(গ) ইউসুফ আলী (ঘ) এ.এইচ.এম. কামরুজ্জামান

২৯। বলাভর কার্যকর হয় কত সালে?

- (ক) ১৮৯৯ (খ) ১৯০৩  
(গ) ১৯০৫ (ঘ) ১৯১১

৩০। সেনরা কোথা থেকে এসেছিল?

- (ক) উত্তর ভারত (খ) দক্ষিণ ভারত  
(গ) পাঞ্জাব (ঘ) সিন্ধু

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with 1200 ml of 2M  $\text{HNO}_3$  if the final density of solution is same as pure water then molarity of the iron in resulting solution which decides the nature of the above solution is 1 0.25M 2 0.5M 3 0.888M 4 none of

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web what is the formula for converting molarity to molality molality is equal to the product of molarity divided by the density of the solution at that temperature  $M \times \frac{1}{d}$  which is more important molality or molarity both molality and molarity are concentration units and they are used interchangeably

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web correct answer c both a and b q2 the density of 1 M solution of HCl is 1.0585 g/ml the molality of the solution is 1.0585 1.010 0.0585 correct answer b 1 explanation molality 
$$m = \frac{\text{moles of solute}}{\text{kg of solvent}}$$

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web sep 13 2023 the only difference between molality and molarity is in the denominator molality is measured in kilograms of solvent whereas molarity is measured in liters of solution when studying the properties of solutions related to vapour pressure and temperature changes concentrations expressed in molality are used

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web molarity is the number of moles of a substance per litre of solution also known as molar concentration a capital M signifies solutions labelled with molar concentration a 1.0 M solution contains 1 mole of solute per litre of solution molality is the number of solvent moles per kilogram

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web 1 molarity  $M = \frac{m}{V}$  mass molar mass  $\times 0.4500 \text{ L} = 0.825 \text{ g}$  141.9579 g/mol  $\times 0.0129 \text{ M} = 2 \text{ molality}$  0.825 g 141.9579 g/mol 0.00581158 mol 0.00581158 mol 0.4500 kg 0.0129 M 3 mole fraction  $n_2 = \frac{0.825 \text{ g}}{141.9579 \text{ g/mol}} = 0.00581158 \text{ mol}$   $n_2 = 0.450 \text{ g} \times 18.015 \text{ g/mol} = 8.10675 \text{ g}$  24.97918401 mol

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web apr 7 2014 molarity moles of solute / liters of solution 8.422 a first convert 250 ml to liters  $250 \times \frac{1}{1000} = 0.25$  then calculate molarity 5 moles / 0.25 liters = 20 M 3 c a solution with molarity 2 requires 2 M of NaOH per liter so  $4 \times 2.8 \text{ M} = 11.2 \text{ M}$

solution of molarity 1.5 M requires 1.5 mol of Na to every litre of solvent

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web answer the correct option is c explanation the molarity of a mixture mix can be calculated using the following formula  

$$M_{\text{mix}} = \frac{M_1 V_1 + M_2 V_2 + \dots + M_n V_n}{V_1 + V_2 + \dots + V_n}$$
 1.5 x 480 + 1.2 x 520 = 480 + 520 = 1000  

$$M_{\text{mix}} = \frac{720 + 624}{1000} = 1.344 \text{ M}$$
 with increase in temperature which of these changes a molality b fraction of solute present in water

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web determine the molarity for each of the following solutions 0.444 mol of  $\text{COCl}_2$  in 0.654 l of solution 98.0 g of phosphoric acid  $\text{H}_3\text{PO}_4$  in 1.00 l of solution 0.2074 g of calcium hydroxide  $\text{Ca(OH)}_2$  in 40.00 ml of solution 10.5 kg

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web molality differs from molarity only in the denominator while molarity is based on the liters of solution molality is based on the kilograms of solvent concentrations expressed in molality are used when studying properties of solutions related to vapor pressure and temperature changes

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