सामान्य निर्देश :

निम्नलिखित निर्देशों को बहुत सावधानी से पढ़िए और उनका सख़्ती से पालन कीजिए :

- (i) यह प्रश्न-पत्र चार खण्डों में विभाजित किया गया है क, ख, ग और घ।
- (ii) इस प्रश्न-पत्र में 37 प्रश्न हैं । सभी प्रश्न अनिवार्य हैं ।
- (iii) खण्ड क प्रश्न संख्या 1 से 20 तक अति लघु-उत्तरीय प्रश्न हैं, प्रत्येक प्रश्न 1 अंक का है।
- (iv) खण्ड ख प्रश्न संख्या 21 से 27 तक लघु-उत्तरीय प्रश्न हैं, प्रत्येक प्रश्न 2 अंकों का है।
- (v) **खण्ड ग** प्रश्न संख्या 28 से 34 तक दीर्घ-उत्तरीय प्रकार के प्रश्न हैं, प्रत्येक प्रश्न 3 अंकों का है।
- (vi) **खण्ड घ** प्रश्न संख्या **35** से **37** तक भी दीर्घ-उत्तरीय प्रकार के प्रश्न हैं, प्रत्येक प्रश्न **5** अंकों का है।
- (vii) प्रश्न-पत्र में कोई समग्र विकल्प नहीं है। तथापि, एक-एक अंक के दो प्रश्नों में, दो-दो अंकों वाले दो प्रश्नों में, तीन-तीन अंकों वाले एक प्रश्न में तथा पाँच-पाँच अंकों वाले तीनों प्रश्नों में आंतरिक विकल्प दिया गया है। ऐसे प्रश्नों में केवल एक ही विकल्प का उत्तर दीजिए।
- (viii) इसके अतिरिक्त, आवश्यकतानुसार, प्रत्येक खण्ड और प्रश्न के साथ यथोचित निर्देश दिए गए हैं।
- (ix) केल्कुलेटरों अथवा लॉग टेबलों के प्रयोग की अनुमति **नहीं** है।
- (x) जहाँ आवश्यक हो, आप निम्नलिखित भौतिक नियतांकों के मानों का उपयोग कर सकते हैं :

$$c = 3 \times 10^8 \text{ m/s}$$
 $h = 6.63 \times 10^{-34} \text{ Js}$
 $e = 1.6 \times 10^{-19} \text{ C}$
 $\mu_0 = 4\pi \times 10^{-7} \text{ T m A}^{-1}$
 $\epsilon_0 = 8.854 \times 10^{-12} \text{ C}^2 \text{ N}^{-1} \text{ m}^{-2}$
 $\frac{1}{4\pi\epsilon_0} = 9 \times 10^9 \text{ N m}^2 \text{ C}^{-2}$
इलेक्ट्रॉन का ट्रब्यमान $(m_e) = 9.1 \times 10^{-31} \text{ kg}$
न्यूट्रॉन का ट्रब्यमान $= 1.675 \times 10^{-27} \text{ kg}$
प्रोटॉन का ट्रब्यमान $= 1.673 \times 10^{-27} \text{ kg}$
आवोगाद्रो संख्या $= 6.023 \times 10^{23} \text{ प्रति ग्राम मोल}$
बोल्ट्रमान नियतांक $= 1.38 \times 10^{-23} \text{ JK}^{-1}$

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Alfred Emanuel Smith, Francis Walton

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