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to find out which metal is used in each of the five unknown salts by using a metal loop placing it in the unknown salts and
testing it by putting the metal under a roaring bunsen flame safety points wear goggles all the time when

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metallurgy and not the nonmetals in this combinations approved which heavy metal had monitored to expend

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web the easiest colors to observe during the flame test were that of copper chloride and lithium chloride for potassium
chloride there was a purple color close to the stick but the flame remained orange for the majority of the burn time the
unknown slat burned orange but also had a slight blue color close to the stick

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identified the element beyond question this method of using light to make identifications became known as spectroscopy

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farbigkeit performing calculations on determine the frequency and energy of the emitted

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web use crayons or colored pencils to color in the spectrum in the box on your report sheet to show the correlation between

color and wavelength part ii flame tests you will conduct flame tests to observe the flame emission colors for the following

solutions lino3 cu no3 2 sr no3 2 ba no3 2 kno3 and nano3 since nitrates do not emit color you

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spectra from red fireworks to a substance in your material list then make your final determination of what ion in the material

list causes red fireworks procedure

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