

BUILD Vocabulary

ionic bond chemical bond formed when one or more electrons are transferred from one atom to another

ion atom that has a positive or negative charge

covalent bond type of bond between atoms in which the electrons are shared

molecule smallest unit of most compounds that displays all the properties of that compound

van der Waals forces weak attraction that develops between oppositely charged regions of nearby molecules

Preface The prefix *co-* and *co-* mean "together." A compound is made up of elements joined together, and a covalent bond connects atoms that share electrons together. **How are covalent bonds different from ionic bonds?**

Chemical Bonds

KEY QUESTION What are the main types of chemical bonds?

The atoms in compounds are held together by chemical bonds. Bonds are formed by the electrons in the outer shell, which are called valence electrons. The main types of chemical bonds are ionic bonds and covalent bonds.

Ionic Bonds **Ionic bonds** are formed when one or more electrons are transferred from one atom to another. This transfer causes the atom losing the electrons to become positively charged. The atom gaining the electrons becomes negatively charged. The positively and negatively charged atoms become **ions**. The oppositely charged ions are attracted to each other and form an ionic bond.

Covalent Bonds Sometimes atoms share electrons rather than transferring them. These shared electrons move around the nuclei of both atoms. The bond formed by sharing electrons is called a **covalent bond**. Atoms can share between two and six electrons. The structure that results is known as a **molecule**. A molecule is the smallest unit of most compounds.

Weak Interactions The strongest chemical bonds are ionic or covalent. Weak interactions between atoms are also important. Within living organisms, many molecules interact only briefly. **Van der Waals forces** are one type of weak interaction between molecules. They occur when molecules are very close together.

Hydrogen bonds are another type of weak interaction. They occur between a hydrogen atom of one molecule and an oxygen or nitrogen atom of a neighboring molecule.

Visual Reading Skill Covalent Bonds

1. What kind of chemical bond holds the water molecule together? _____
2. What particles are shared between the atoms in the molecule? _____
3. What is the other main type of chemical bond? _____
4. What is the difference between covalent and ionic bonds? _____



Water molecule (H_2O)

Properties of Water

READING TOOL **Cause and Effect** As you read the part of the lesson under "The Water Molecule," use the table below to list the causes and effects of the properties of water. Many of the causes may be the same, but each characteristic will have a different effect. Part of the table is completed for you.

Characteristic of Water	Cause	Effect
Hydrogen Bonding		The slightly negative end of a water molecule is attracted to the slightly positive end of another water molecule.
Expands Upon Freezing		
Dissolves Many Substances	Water forms hydrogen bonds.	
Cohesion		
Adhesion		
Heat Capacity		

Miller And Levine Biology Workbook Chapter 7 Key

Richard Bailey



Miller And Levine Biology Workbook Chapter 7 Key:

Nunn's Applied Respiratory Physiology E-Book Andrew B. Lumb, 2012-09-25 Nunn's Applied Respiratory Physiology Seventh Edition covers all aspects of respiratory physiology in health disease and altered conditions and environments from basic science to clinical applications Includes functional anatomy mechanics control of breathing ventilation circulation ventilation perfusion matching diffusion carbon dioxide and oxygen and non respiratory functions of the lung Discusses the effects of pregnancy exercise sleep altitude pressure drowning smoking anaesthesia hypocapnia hypercarbia hypoxia hyperoxia and anaemia on respiratory physiology Explores specific clinical disorders such as ventilatory failure airways disease pulmonary vascular disease parenchymal lung disease and acute lung injury as well as the physiological basis of current therapies including artificial ventilation extrapulmonary gas exchange and lung transplantation Chapter on Parenchymal Lung Disease has been specifically expanded to include the physiology and pathology of the pleural space and lung cancer Contains a new chapter on Pulmonary Surgery covering a wide range of surgical interventions from bronchoscopy to lung resection Includes almost 500 new references to the literature The result is an invaluable source for those preparing for examinations in anaesthesia and intensive care as well as an essential purchase for practitioners who want quick reference to current knowledge Describes respiration in health and disease and in normal and abnormal situations to help readers manage all conditions they see in their practices Examines the respiratory effects of exercise sleep smoking anaesthesia drowning anaemia pregnancy and other events as well as environmental factors such as altitude flying high pressure closed environments and air pollution on respiration Maintains the clarity of style and single author approach of previous editions through the close collaboration of Andrew Lumb and John Nunn Makes difficult concepts easy to understand and apply with nearly 300 illustrations A new chapter on the History of Respiratory Physiology More coverage of pathophysiology and even more applications of respiratory physiology to clinical practice A more consistent organization a revised page design that aids readability and an art program featuring new and newly redrawn illustrations Biology

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with a page reference to the corresponding Nursing Management Plan Nursing Management Plans provide you with a complete care plan for every Priority Diagnosis that includes the diagnosis definition defining characteristics outcome criteria nursing interventions and rationales Evidence Based Collaborative Practice boxes summarize evidence based recommendations for a variety of therapies Collaborative Management boxes guide you through the management of a wide variety of disorders Patient Education boxes list the concepts that must be taught to the patient and the family before discharge from the ICU Concept maps help you understand common critical health conditions including acute coronary syndrome acute renal failure ischemic stroke and shock NEW Case studies with critical thinking questions test your understanding of key concepts and their practical applications NEW Priority Medication boxes give you a foundation in the pharmacology used most in critical care UPDATED New information on the management of the alcoholic patient and disorders resulting from alcoholism is added to chapter nine *Gray's Anatomy E-Book* Susan Standring, 2021-05-22

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