

# PRACTICE PAPER 2024 JEE MAIN

Jan-24

## SECTION-A (MULTIPLE CHOICE QUESTIONS)

1. Read the following statements and choose the correct option.

- (i) Amines can be distinguished by Hinsberg's reagent.  
(ii) The three types of amines react similarly with Hinsberg's reagent and hence can be distinguished from each other.
- (a) (i) is true and (ii) is true.  
(b) (i) is true and (ii) is false.  
(c) (i) is false and (ii) is true.  
(d) (i) is false and (ii) is false.

2. One mole of an organic compound A with the formula  $C_3H_8O$  reacts completely with two moles of HI to form X and Y. When Y is boiled with aqueous alkali, it forms Z. Z answers the iodoform test. The compound A is

- (a) propan-2-ol (b) propan-1-ol  
(c) ethoxyethane (d) methoxyethane.

3. 0.30 g of an organic compound containing C, H and O on combustion yields 0.44 g  $CO_2$  and 0.18 g  $H_2O$ . If one mol of compound weighs 60, then molecular formula of the compound is

- (a)  $C_3H_8O$  (b)  $C_2H_4O_2$   
(c)  $CH_2O$  (d)  $C_4H_8O$

4. The correct order of hybridisation of the central atom in the following species  $NH_3$ ,  $[PtCl_4]^{2-}$ ,  $PCl_5$  and  $BCl_3$  is

- (a)  $dsp^2$ ,  $dsp^2$ ,  $sp^2$ ,  $sp^3$  (b)  $sp^3$ ,  $dsp^2$ ,  $sp^3$ ,  $d$ ,  $sp^2$   
(c)  $dsp^2$ ,  $sp^2$ ,  $sp^3$ ,  $dsp^2$  (d)  $dsp^2$ ,  $sp^2$ ,  $sp^2$ ,  $dsp^2$

5. Arrange the following compounds in increasing order of oxidation number.

$MnCl_2$ ,  $MnO_2$ ,  $Mn(OH)_3$ ,  $KMnO_4$

- (a)  $MnCl_2 < MnO_2 < Mn(OH)_3 < KMnO_4$   
(b)  $MnO_2 < MnCl_2 < KMnO_4 < Mn(OH)_3$   
(c)  $Mn(OH)_3 < MnCl_2 < MnO_2 < KMnO_4$   
(d)  $MnCl_2 < Mn(OH)_3 < MnO_2 < KMnO_4$

6. Elements X, Y and Z have atomic numbers 19, 37 and 55 respectively. Which of the following statements is true about them?

- (a) Their ionisation potential would increase with increasing atomic number.  
(b) Y would have an ionisation potential between those of X and Z.  
(c) Z would have the highest ionisation potential.  
(d) Y would have the highest ionisation potential.

7. Statement-1 : Bromobenzene upon reaction with  $Br_2/Fe$  gives 1, 4-dibromobenzene as the major product.

Statement-2 : In bromobenzene, the inductive effect of the bromo group is more dominant than the mesomeric effect in directing the incoming electrophile.

- (a) Statement-1 is true; statement-2 is true; statement-2 is a correct explanation for statement-1.  
(b) Statement-1 is true; statement-2 is true; statement-2 is not a correct explanation for statement-1.  
(c) Statement-1 is true, statement-2 is false.  
(d) Statement-1 is false, statement-2 is true.

8. What would be the wavelength and name of series respectively for the emission transition for H-atom if it starts from the orbit having radius 1.3225 nm and ends at 211.6 pm?

- (a) 434 nm, Balmer (b) 434 pm, Paschen  
(c) 545 pm, Pfund (d) 600 nm, Lyman

9. Consider the following statements :

- (I)  $CH_3\overset{\cdot}{O}\overset{\cdot}{C}H_2$  is more stable than  $CH_3\overset{\cdot}{C}H_2$ .  
(II)  $Me_2\overset{\cdot}{C}H$  is more stable than  $CH_3CH_2\overset{\cdot}{C}H_2$ .  
(III)  $CH_2 = \overset{\cdot}{C}H - \overset{\cdot}{C}H_2$  is more stable than  $CH_3CH_2\overset{\cdot}{C}H_2$ .  
(IV)  $CH_2 = \overset{\cdot}{C}H$  is more stable than  $CH_3\overset{\cdot}{C}H_2$ .

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**Puja Mehta**



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