

Q1

The common oxidation states of Ti are

- (a) +2 and +3 (b) +3 and +4  
(c) -3 and -4 (d) +2, +3 and +4

Q2

When phenol is treated with excess bromine water. It gives

- (a) *m*-Bromophenol (b) *o*- and *p*-Bromophenols  
(c) 2,4-Dibromophenol (d) 2,4,6-Tribromophenol.

Q3

If 0.5 mol of  $\text{BaCl}_2$  is mixed with 0.2 mole of  $\text{Na}_3\text{PO}_4$ , find the maximum amount of  $\text{Ba}_3(\text{PO}_4)_2$  that can be formed.

- (a) 1 mole (b) 0.5 mole (c) 0.1 mole (d) 0.01 mole

Q4

Which is chemically most active non-metal?

- (a) S (b) O (c) F (d) N

Q5

The following species will not exhibit disproportionation reaction

- (a)  $\text{ClO}^-$  (b)  $\text{ClO}_2^-$  (c)  $\text{ClO}_3^-$  (d)  $\text{ClO}_4^-$

Q6

Which of the following has been used in the manufacture of non-inflammable photographic films?

- (a) Cellulose nitrate (b) Cellulose xanthate  
(c) Cellulose perchlorate (d) Cellulose acetate

Q7

Point out the wrong statement:

For a first order reaction

- (a) time for half-change ( $t_{1/2}$ ) is independent of initial concentration  
(b) change in the concentration unit does not change the rate constant ( $k$ )  
(c) time for half-change  $\times$  rate constant = 0.693  
(d) the unit of  $k$  is  $\text{mol}^{-1} \text{min}^{-1}$

Q8

Name of the structure of silicates in which three oxygen atoms of  $[\text{SiO}_4]^{4-}$  are shared.

- (a) Pyrosilicate  
(b) Sheet silicate  
(c) Linear chain silicate  
(d) Three dimensional silicate

Q9

In the following sequence of reactions



- (a) *n*-propyl amine (b) isopropylamine  
(c) ethylamine (d) ethylmethyl amine

Q10

Which of the following monosaccharide is a pentose?

- (a) Galactose (b) Glucose  
(c) Fructose (d) Arabinose

Q11

In Bohr's model, atomic radius of the first orbit is  $y$ , the radius of the 3<sup>rd</sup> orbit, is

- (a)  $y/3$  (b)  $y$   
(c)  $3y$  (d)  $9y$

Q12

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