

PHYSICAL: Kinetic Molecular Theory

Learning Objectives

- to identify/illustrate differences in how gases, liquids/solids, and plasmas behave; to describe and understand the properties of the kinetic molecular theory;
- to identify/illustrate differences in the properties of matter;
- to represent quantitative values using scientific notation and SI units.

Module 1: Representation of Matter in Different Phases

The following figure illustrates the representation of particles in three different phases of matter:



Figure 1: Representations of particles in different phases of matter

Students might be confused by the different representations of particles in different phases of matter. It is important to note that the representation of particles in different phases of matter is not a perfect representation of reality.

Key Concepts

- Effect: The key concept of the kinetic molecular theory is that matter is made of particles, which are in constant motion. The particles are in constant motion in all phases of matter. The particles are in constant motion in all phases of matter. The particles are in constant motion in all phases of matter.

Table 1: Kinetic Molecular Theory of Matter

	GAS	LIQUID	SOLID
How are the particles arranged?	Particles are in a random arrangement. They are moving in all directions. The particles are in a random arrangement.	Particles are in a random arrangement. They are moving in all directions. The particles are in a random arrangement.	Particles are in a regular arrangement. They are moving in all directions. The particles are in a regular arrangement.
How are the particles moving?	Particles are in a random arrangement. They are moving in all directions. The particles are in a random arrangement.	Particles are in a random arrangement. They are moving in all directions. The particles are in a random arrangement.	Particles are in a regular arrangement. They are moving in all directions. The particles are in a regular arrangement.

Kinetic Molecular Theory Pogil Instructors Guide

Eric Hutchinson



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Teachers.Net Lesson Exchange: Kinetic Molecular Theory with Popcorn , Cec Knight offers a lesson for middle and high school students that uses popcorn to demonstrate the kinetic molecular theory Knight includes a list of the materials required the time needed and the procedures Teachers Net provides the lesson as part of the Teachers Net Lesson Exchange online resource *Kinetic Molecular Theory* Charles Proteus Steinmetz,American Electrochemical Society. Meeting,1908* **Instructor's Guide for Experiments in General Chemistry** ,1996 Instructor's Guide for Experiments in General Chemistry Steven Murov,1996 **Communities of Molecules** Howard DeVoe,Robert Hearle,1978 Instructor's Guide for Dickerson/Gray/Haight Chemical Principles Ben B. Chastain,Richard Earl Dickerson,Harry B. Gray,Gilbert Pierce Haight,1979 Chemical Principles Emil J. Slowinski,1973 **Instructor's Guide for Chemical Principles** Richard Earl Dickerson,1984 Instructor's Guide to Chemistry for Changing Times John William Hill,1984 Chemistry Steven S. Zumdahl,Susan A. Zumdahl,Donald J. DeCoste,2007 **An Instructor's Guide to Physical Chemistry** Eric Hutchinson,19?? Instructor's Guide to Problems in Principles of Chemistry Paul Ander,Anthony J. Sonnessa,1965 *Introductory Chemistry* Steven S. Zumdahl,1990-06-01 *Instructor's Guide to Experimental General Chemistry* William Thomas Lippincott,Devon W. Meek,Frank Henry Verhoek,1984 Instructor's Resource Guide to Chemistry Steven S. Zumdahl,1997 **Physics** Ira Maximilian Freeman,1968 *Instructor's Guide to the Problems in Physical Chemistry* John A. Ricketts,1961 **Instructor's Guide for Chemistry for Changing Times, Second Edition** John William Hill,1975 General Chemistry: an Integrated Approach Petrucci Hill,2004-07 *Kinetic Theory* ,19??

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