

KINETIC THEORY OF MATTER WORKSHEET

1. A student performed a simple experiment using liquid bromine. She placed a small amount in the bottom of a sealed gas jar of air. After two minutes, red-brown fumes were seen just above the liquid bromine surface. After one hour, the red-brown fumes had spread completely throughout the gas jar.



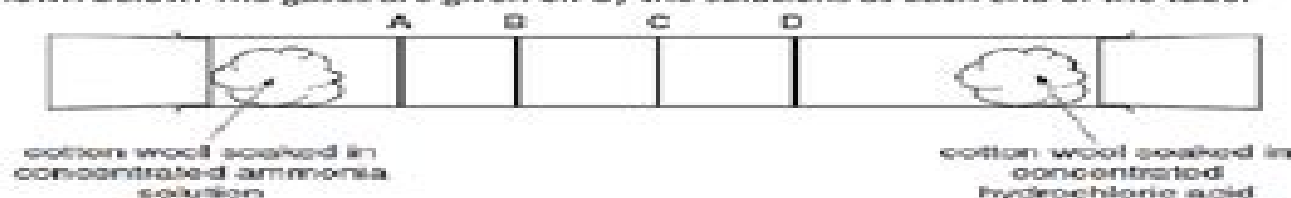
Explain these observations using the kinetic theory of matter

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2. The diffusion of ammonia gas and hydrogen chloride gas in a glass tube are shown below. The gases are given off by the solutions at each end of the tube.



- a. What is formed when ammonia gas mixes with hydrogen chloride gas?
-
- b. Which line shows where the product will be formed?
-
- c. Give a reason for your answer in b above using Graham's law of diffusion
-
- d. Is the product formed a solid, liquid or gas?
-

Kinetic Theory Section Reinforcement Answers

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