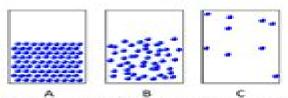
Kinetic Molecular Theory Worksheet

Use the following terms to fill in the blanks. Some may be uses more than once. Hopefully all will be included but you never know. If you seem to need an additional term (oops!) just do your best

Gas	Solid	Liquid	Phase	Compress
Empty Space	Kinetic Energy	Bonds	Structure	Collisions
Anomaly	Random Motion	Temperature	Absolute Zero	density
Fluids	Compressible	Incompressible.		



A	SOLID LIQUID GAS	
В		
С		

The boxes above represent matter in the 3 common phases. Write the name of the phase next to the appropriate letter.

1. The word kinetic comes from a Greek word that means "to move." The kinetic molecular theory is

based upon the assumption that atoms are in constant Random Motion

State the names of the phases in order of Temperature, Kinetic Energy, density and bond strength between atoms

Lowest Temperatures	In between Temperatures	Highest Temperatures
Solid	Liquid	Gas
Lowest Kinetic Energy	In between Kinetic Energy	Highest Kinetic Energy
Solid	Liquid	Gas
Lowest Density		Highest Density
Gas		
High Strength Bonds	Weak Bonds	No Bonds
Solid	Liquid	Gas

- 3. Kinetic Energy is directly related to temperature. As the temperature increases so does the Kinetic Energy.
- Compared to Solids and Liquids gases tend to have a lower density. This can be explained by the fact that there is so much Empty Space between particles in a gas.
- A small pressure will easily compress a gas. If this happens its density will increase. The space between the atoms will (INCREASE or DECREASE).
- Even a large pressure will not compress a solid because the atoms are so closely packed.
- Solids have strong Bonds, liquids have weaker Bonds and gases have no Bonds holding them together.
- A solid like ice or diamond has a highly ordered arrangement of bonds between atoms which is called its Structure. A liquid or gas does not have this.

Kinetic Molecular Theory Work Sheet Answer Key Pogil

Amelia E. Punzalan

Kinetic Molecular Theory Work Sheet Answer Key Pogil:

Kinetic Molecular Theory Charles Proteus Steinmetz, American Electrochemical Society. Meeting, 1908*

Teachers.Net Lesson Exchange: Kinetic Molecular Theory with Popcorn, Cec Knight offers a lesson for middle and high school students that uses popcorn to demonstrate the kinetic molecular theory Knight includes a list of the materials required the time needed and the procedures Teachers Net provides the lesson as part of the Teachers Net Lesson Exchange online resource

A Content Analysis of Kinetic Molecular Theory Laboratory Activities in Middle School Physical Science Textbooks Janet Elizabeth Thompson, 2001

Student's Use of the Kinetic Theory to Explain Familiar Events

Amelia E. Punzalan, 1985

Molecular Physics: Kinetic Theory and Thermodynamics S. K. Sinha, T.K. Dey, 2006

Molecules and Motion Jon Ogborn, 1973

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