

AS-Chapter 3 Handout

1. (a) Sodium reacts with cold water

(b) What would you see as the reaction proceeds?

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(c) Write the balanced chemical equation for this reaction

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(d) Calculate the volume of gas produced if 3.0 g of sodium reacts with an excess of water

(One mole of any gas at the temperature and pressure of the experiment occupies  $24 \text{ dm}^3$ )

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Total 7 marks

2. (a) (i) What is meant by the mass number of an atom?

.....

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# Leaked May 2014 Ial Paper 3 Chemistry

**Judd E. Hollander**



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