

# Chemistry 2 Student Workbook

## Answers to Worksheets

Teacher Notes

### UNIT 3

#### Worksheet 01 Crossword—exploring energy

##### ACROSS

- 3 Coal seam  
5 Enthalpy  
7 Exothermic  
11 Cathode  
12 Renewable  
14 Thermochemical  
18 Salt bridge  
19 Secondary  
21 Hygroscopic  
22 Fuel  
23 Anode

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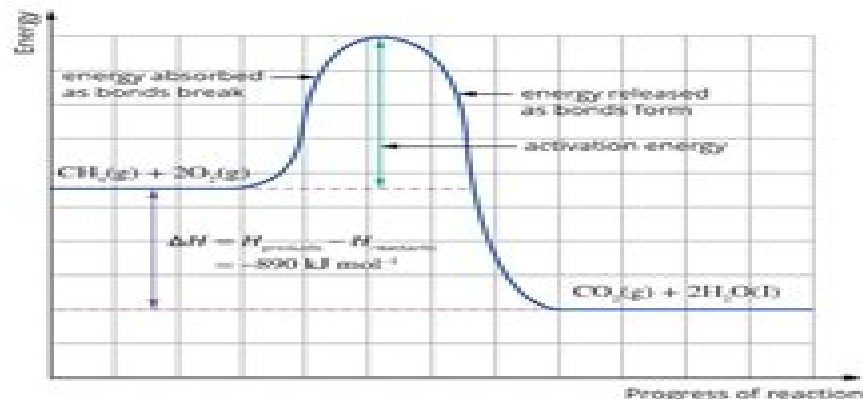
- 1 Galvanic  
2 Biofuel  
4 Spontaneous  
6 Biogas  
8 Joule  
9 Oxidation  
10 Greenhouse  
13 Activation  
15 Electrolyte  
16 Viscosity  
17 Reduction  
20 Fossil  
22 Fuel

#### Worksheet 02 Combustion of fuels—thermochemical equations and specific heat capacity

- 1 The combustion of fuels releases a significant quantity of energy that can be used as an **energy source**. **Chemical energy** is the energy stored in the bonds between atoms and between molecules arising from attractions and repulsions between subatomic particles. It is called **enthalpy** and is represented by the symbol, ***H***. All combustion reactions are **exothermic** reactions. In such reactions, the enthalpy change, symbol  **$\Delta H$** , has a **negative** sign.

$$\Delta H = H_{\text{products}} - H_{\text{reactants}}$$

2



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**Karin Nielsen-Saines**



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## **Table of Contents Lesson 8 2 Workbook Answers Pearson Education Chemistry**

1. Understanding the eBook Lesson 8 2 Workbook Answers Pearson Education Chemistry
  - The Rise of Digital Reading Lesson 8 2 Workbook Answers Pearson Education Chemistry
  - Advantages of eBooks Over Traditional Books
2. Identifying Lesson 8 2 Workbook Answers Pearson Education Chemistry
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Lesson 8 2 Workbook Answers Pearson Education Chemistry
  - User-Friendly Interface
4. Exploring eBook Recommendations from Lesson 8 2 Workbook Answers Pearson Education Chemistry
  - Personalized Recommendations
  - Lesson 8 2 Workbook Answers Pearson Education Chemistry User Reviews and Ratings
  - Lesson 8 2 Workbook Answers Pearson Education Chemistry and Bestseller Lists
5. Accessing Lesson 8 2 Workbook Answers Pearson Education Chemistry Free and Paid eBooks
  - Lesson 8 2 Workbook Answers Pearson Education Chemistry Public Domain eBooks
  - Lesson 8 2 Workbook Answers Pearson Education Chemistry eBook Subscription Services
  - Lesson 8 2 Workbook Answers Pearson Education Chemistry Budget-Friendly Options
6. Navigating Lesson 8 2 Workbook Answers Pearson Education Chemistry eBook Formats

- ePub, PDF, MOBI, and More
  - Lesson 8 2 Workbook Answers Pearson Education Chemistry Compatibility with Devices
  - Lesson 8 2 Workbook Answers Pearson Education Chemistry Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Lesson 8 2 Workbook Answers Pearson Education Chemistry
  - Highlighting and Note-Taking Lesson 8 2 Workbook Answers Pearson Education Chemistry
  - Interactive Elements Lesson 8 2 Workbook Answers Pearson Education Chemistry
8. Staying Engaged with Lesson 8 2 Workbook Answers Pearson Education Chemistry
- Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Lesson 8 2 Workbook Answers Pearson Education Chemistry
9. Balancing eBooks and Physical Books Lesson 8 2 Workbook Answers Pearson Education Chemistry
- Benefits of a Digital Library
  - Creating a Diverse Reading Collection Lesson 8 2 Workbook Answers Pearson Education Chemistry
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Lesson 8 2 Workbook Answers Pearson Education Chemistry
- Setting Reading Goals Lesson 8 2 Workbook Answers Pearson Education Chemistry
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Lesson 8 2 Workbook Answers Pearson Education Chemistry
- Fact-Checking eBook Content of Lesson 8 2 Workbook Answers Pearson Education Chemistry
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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