Score: _____ Date:

Practicing with Isotopes

1. Complete the following questions. Assume all the atoms are neutral.

⁴He

56 Fe

27 AI

Element : # of protons: Element: # of protons: Element: # of protons:

of neutrons:

of neutrons:

of neutrons:

40 Ca

14 Ca

Element

Element:

Element: # of protons:

of protons: # of neutrons:

of protons: # of neutrons:

of neutrons:

 ^{1}H

 ^{2}H

Element:

Element:

Element:

of protons:

of protons:

of protons:

of neutrons:

of neutrons:

of neutrons:

2. Calculate the average atomic mass for the following element from the given information about the relevant isotopes:

Isotope 1: Mass = 35 amu

Abundance = 75.53%

Isotope 2: Mass = 37 amu

Abundance = 24.47%

Isotopes Practice Set Key

Ensheng Dong

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