AICE Marine Science Semester 1 Review

Unit 1 - Water

Key Vocabulary:

- Kinetic particle theory
 Water vapour
 Proton
 Neutron
 Electron
 Nucleus
 Shells
 Atomic number
 Bond
 Emergent properties
 Covalent bond
 Ionic bond
 Hydrogen bond
- Polar
 Solvent
 Density
 Specific heat capacity
 Solute
 Solution
 Solubility
 Dissolution
 Salinity
 Precipitation
 Run-off
 Evaporation
 Hypersaline
 Halocline

Thermocline
PH
Acidic
Alkaline
Neutral
PH scale
Turbulence
Dissolved oxygen
Phytoplankton
Oxygen minimum layer
Water column
Gradient
Thermal insulator
Pycnocline

- 1. List the 5 oceans and know their locations
- Identify the 4 main layers/zones of the ocean and which ones are photic?
- Identify when you would use the different words: pelagic and benthic?
- 4. Explain the kinetic particle theory of WATER and the differences between solids, liquids and gases.
- 5. Identify the following items found in the Ocean
 - a. Key elements
 - b. Key compounds (ionic and covalent)
 - c. Key mixtures
- Draw the structure of the following atoms (include nucleus, protons, neutrons, and electrons (show arrangement of electrons and shells/rings))
 - a. Oxygen
 - b. Hydrogen
 - c. Carbon
 - d. Sodium
 - e. Chlorine-35
- Using a diagram describe the covalent bonding in a water molecule including the sharing of electron pairs.

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ML Morrison

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