## Mark scheme Chemistry Past Paper Questions 1

#### SECTION B: CHEMISTRY

Item:	Expected answers	Additional guidance	Marks
4 (a)	5		1
4 (b)	An electron has a much smaller mass than a proton.		7
4 (c)	(20 x 10) + (80 x 11)/100 (1) = 10.8 (1)	(200 + 80)/100 (1) or 1080/100 (1)	2
4 (d)	2 or 8 but not 3 for 1 mark 2, 8, 3 for 2 marks		2
		Total mark	6

Item	Expected answers	Additional guidance	Marks
5 (a)	Any one of the following: It is on the right hand side of the periodic table (1) It is an inert gas/rare gas/noble gas (1)		1
They have similar other (1)		Do not accept 'they are all alkali metals'	2
	They have similar properties to each other (1)		
	They all have 1 electron in their outer shell (1)		
		Total mark	3

# **Mark Scheme Chemistry June 2013 Paper 2**

**Mark Steyvers** 

 $\label{lem:mark-scheme} \textbf{Mark Scheme Chemistry June 2013 Paper 2:}$ 

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