

# GUIDA PER L'UTENTE



# Manuale Istruzioni Samsung Galaxy Gio

**Lara Lazzeroni**



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finite size and volume the collisions between the ideal gas particles are said to be elastic they exert no attractive or repulsive forces hydrogen gas generated in today s experiment is however a real gas not an ideal gas

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web possible answer butane is extremely flammable and will ignite possible answer because the butane will not dissolve in the water it is possible for butane gas possible to form possible answers 0 c 273 k and 1 atm 101 kpa c 12 g mol 4 48 g mol h 1 g mol 10 10 g mol mm 48 g mol 10 g mol 58 g mol

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web use the combined gas law to determine the volume in l of butane at stp use avogadro s law to determine the number of moles of butane gas assume that butane is an ideal gas and that one mole has a volume of 22 l at stp experimental molar mass

g mole of butane the molecular formula of butane is  $C_4H_{10}$

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web although the most convenient gas for this is butane other gases may be available in similar small cylinders the collection of a gas sample and the weighing of the gas cylinder before and after this should take about 5 10 minutes as a demonstration