

**Program: B.Tech**
**Course Code & Title: 21CVB101J/Chemistry**
**Year & Sem: I/II**
**Date: 29-03-2023**
**Time: 12.30-1.30 pm**
**Max. Marks: 30 Marks**
**Answer ALL the MCQs Part-A (10\*1=10 Marks)**

- Which of the following is not a state function  
 a)  $\Delta G$    b)  $\Delta E$    c)  $W$    d)  $H$
- Internal energy does not include  
 a) Nuclear energy   b) Vibrational energy   c) Rotational energy   d) Energy of gravitational pull
- Gibbs energy change  $\Delta G$  is related to equilibrium constant  $K$  as:  
 a)  $\Delta G^\circ = -RT \ln k$    b)  $\Delta G^\circ = RT \ln k$    c)  $\ln k = -RT/\Delta G^\circ$    d)  $\ln k = \Delta G^\circ/RT$
- The conductivity of electrolytic conductors is due to \_\_\_\_\_  
 a) Flow of free mobile electrons   b) Movement of ions   c) Either movement of electrons or ions   d) Cannot be said
- The process of deterioration of a metal due to unwanted chemical or electrochemical interaction of the metal with its environment is called \_\_\_\_\_  
 a) Electrolysis   b) Electrodialysis   c) Corrosion   d) Deposition
- Chemical used in salt bridge is  
 a)  $KOH$    b)  $KCl$    c)  $KNO_3$    d)  $KBr$
- If a compound has 3 chiral carbons What is the number of optically active isomers?  
 a) 9   b) 3   c) 4   d) 8
- The potential energy of n-butane is minimum for  
 a) Skew conformations   b) Staggered conformations   c) Eclipsed conformations  
 d) Gauche
- Optical isomerism is a type of \_\_\_\_\_  
 a) metamerism   b) stereoisomerism   c) geometrical isomerism   d) tautomerism

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