

# JEE Main 2020 Paper



**Sol. 2**



P - OH bonds = 4

P = O bonds = 2

P - O - P linkage = 1

Ans. 4, 2, 1

**21.** The mole fraction of glucose ( $\text{C}_6\text{H}_{12}\text{O}_6$ ) in an aqueous binary solution is 0.1. The mass percentage of water in it, to the nearest integer, is \_\_\_\_\_.

**Sol. 47 %**

$$x_{\text{Glucose}} = 0.1$$

$$\begin{aligned} \text{mass\% of glucose} &= \frac{0.1 \times 180}{0.1 \times 180 + 0.9 \times 18} \times 100 \\ &= \frac{1800}{18 + 16.2} \\ &= \frac{1800}{34.2} \\ &= 52.63\% \\ &= 53\% \end{aligned}$$

$$\therefore \text{mass \% of H}_2\text{O} = 47\%$$

**22.** The volume strength of 8.9 M  $\text{H}_2\text{O}_2$  solution calculated at 273 K and 1 atm is \_\_\_\_\_. ( $R = 0.0821 \text{ L atm K}^{-1} \text{ mol}^{-1}$ ) (rounded off to the nearest integer)

**Sol. 100**

$$\begin{aligned} \text{Vol. strength} &= \frac{8.9}{2} \times \frac{0.0821 \times 273}{1} \\ &= 99.73 \\ &= 100 \end{aligned}$$

**23.** An element with molar mass  $2.7 \times 10^{-2} \text{ kg mol}^{-1}$  forms a cubic unit cell with edge length 405 pm. If its density is  $2.7 \times 10^3 \text{ kg m}^{-3}$ , the radius of the element is approximately \_\_\_\_\_  $\times 10^{-12} \text{ m}$  (to the nearest integer).

**Sol. 143**

$$\text{Density} = \frac{Z \times \text{GMM}}{N_A \times a^3}$$

$$\begin{aligned} 2.7 \times 10^3 &= \frac{Z \times 2.7 \times 10^{-2}}{6.023 \times 10^{23} \times (405 \times 10^{-12})^3} \\ Z &= 6.023 \times 405 \times 405 \times 405 \times 10^{23-36+3+2} \\ Z &= 6.023 \times 405 \times 405 \times 405 \times 10^{-8} \\ Z &= 4 \end{aligned}$$

FCC

$$4R = \sqrt{2} \times a$$

$$R = \frac{405}{2\sqrt{2}} \times 10^{-12} = 143.21 \times 10^{-12} \text{ m}$$

= 143 ans

# Jee Main Paper In 6 4 2014

**Oswaal Editorial Board**



## **Jee Main Paper In 6 4 2014:**

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The Union Public Service Commission (UPSC) every year conducts a CDS exam twice a year for candidates who wish to make their career in the defence forces: Army, Navy, and Air Force. The Combined Defence Services Examination is conducted for admission to the Indian Military Academy (IMA), Indian Naval Academy (INA), Air Force Academy (AFA), and Officers Training Academy (OTA). The CDS selection process comprises two stages: written exams and SSB interviews. The final selection of candidates is done based on the performance in both stages. After completing training at IMA, INA, AFA, and OTA, candidates are selected for the post of Lieutenant. In 2024, approx. 4.5 Lacs students applied for the

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