Chemistry Final Exam Review

Multiple Choice

Identify the letter of the	e choice that best comp	detes the statement	or answers the question.
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	1.3	The study of chemicals that, in general, do not contain carbon is traditionally called what type of chemistry?				
		a. bio	c.	physical		
		b. inorganie	d.	analytical		
2.	Which of the following is NOT an example of matter?					
		a. air	C.	smoke		
		b. heat	d.	water vapor		
3.	3.	A golf ball has more mass than a tennis ball because it				
		a. takes up more space	C.	contains different kinds of matter		
		 b. contains more matter 	dl.	has a definite composition		
4.	An example of an extensive property of matter is					
		a. temperature	c.	mass		
		b. pressure	d.	hardness		
5.		All of the following are physical properties of matter EXCEPT				
		a. mass	C.			
		b. color	d.	ability to rust		
	6.	Which state of matter has a definite volum	ne and tak	ces the shape of its container?		
	0.00	a. solid	c.	ons		
		b. liquid	d.	both b and c		
7	Which of the following CANNOT be classified as a substance?					
		a. table salt	C.	nitrogen		
		b. air	d.	gold		
8.	Which of the following is a heterogeneous mixture?					
		a, air	· C.	steel		
		b. salt water	d.	soil		
9.	Which of the following is a heterogeneous mixture?					
		a. vinegar in water	c.	oil and vinegar		
		b. milk	d.	air		
	10.	Separating a solid from a liquid by evaporating the liquid is called				
		a. filtration	C.	solution		
		b. condensation	d.	distillation		
111.		A substance that can be separated into two or more substances only by a chemical change is a(n)				
		a. solution	e.	mixture		
		b. element	d.	compound		
12.		The chemical symbol for iron is				
		a. fe	C.	Fe		
		b. <i>FE</i>	d.	Ir		

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Tobias Bleicker

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