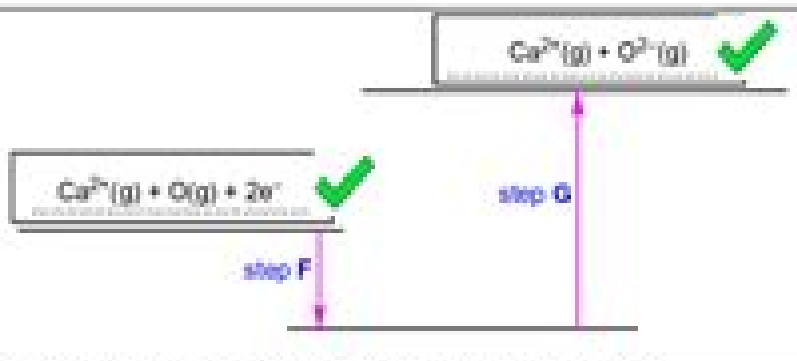


Question	Answer	Marks	Guidance
1 (a)	(The enthalpy change that accompanies) the formation of one mole of a(n ionic) compound ✓ from its gaseous ions (under standard conditions) ✓	2	IGNORE 'energy needed' OR 'energy required' ALLOW as alternative for compound: lattice, crystal, substance, solid Note: 1st mark requires 1 mole 2nd mark requires gaseous ions IF candidate response has '1 mole of gaseous ions', award 2nd mark but NOT 1st mark
(b) (i)		2	Correct species AND state symbols required for both marks $2e^-$ required for left-hand response ALLOW e for e^- Mark each marking point independently
(ii)	(enthalpy change of) formation (of calcium oxide) ✓ (enthalpy change of) atomisation of oxygen ✓ Second electron affinity (of oxygen) ✓	3	calcium oxide not required for this mark DO NOT ALLOW 'lattice formation' (confusion with LE) atomisation AND oxygen/ O_2 / $\frac{1}{2}O_2$ / O both required (atomisation of calcium is also in cycle) IGNORE oxygen or oxygen species, e.g. O^- DO NOT ALLOW calcium

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