

Released Test Questions

Math

6

- 1 Which point shows the location of $\frac{3}{2}$ on the number line?



- A point A
B point B
C point C
D point D

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- 2 Which list of numbers is ordered from *least to greatest*?

- A $\frac{1}{2}$, $2\frac{1}{2}$, 0.2, 0.02
B 0.02, 0.2, $2\frac{1}{2}$, $\frac{1}{2}$
C 0.02, 0.2, $\frac{1}{2}$, $2\frac{1}{2}$
D 0.2, $\frac{1}{2}$, 0.02, $2\frac{1}{2}$

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- 3 Which of the following fractions is closest to 0?

- A $-\frac{5}{12}$
B $-\frac{2}{3}$
C $\frac{5}{6}$
D $\frac{3}{4}$

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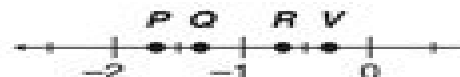
- 4 What is the order of the following set of numbers from greatest to least?

$$1\frac{1}{3}, \frac{2}{3}, -\frac{1}{3}, \frac{9}{5}$$

- A $-\frac{1}{3}, \frac{2}{3}, \frac{9}{5}, 1\frac{1}{3}$
B $-\frac{1}{3}, 1\frac{1}{3}, \frac{2}{3}, \frac{9}{5}$
C $\frac{9}{5}, \frac{2}{3}, 1\frac{1}{3}, -\frac{1}{3}$
D $\frac{9}{5}, 1\frac{1}{3}, \frac{2}{3}, -\frac{1}{3}$

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- 5 Which point *best* represents $-1\frac{3}{8}$ on the number line below?



- A P
B Q
C R
D V

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Math Practice Test 6th Grade 2013

Joseph Prinzevalli



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strategies for mastering every question type 8 week ACT Training Schedule plus last minute Emergency Plan

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given that the boiling points of ethanol and methanol 78 c and 64 c respectively are far off in a significant way this seems like a valid conclusion conclusion thermometer calibrations are necessary in order to collect reliable information as precise

calculations are imperative to gathering correct results

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melting point 1 a push the open end of a capillary tube into the powdered acetamide b move the powder to the closed end of the capillary tube by tapping it on the table repeat until the the powdered acetamide occupies 1 2 mm of the capillary tube end c

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discussion and conclusion the technique recrystallization was used because it is a simple method to purify a solid compound the percentage yield for benzoic acid and acetanilide acid being high indicated the initial sample had a low concentration of impurities with the melting point being in a specific

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conclusion after completing this experiment the melting point of three different unknown compounds was obtained using the mel temp apparatus compound a had a melting point range of 114 6 115 5 c thus revealing it to be acetanilide compound b had a melting point range of 122 6 124 5 c thus

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