

Fill in the table on the right from your periodic table.

Element	Chemical Symbol	Atomic #	Valence Electrons	Oxidation #	# of Protons	Atomic Mass	Metal or Nonmetal	Group or Luster Electrons	# of Full Electron Levels
Oxygen									
Argon									
Calcium									
Nitrogen									
Potassium									
Sulfur									
Helium									

Find the molar mass of the following compounds

Calcium oxide (CaO)

Water

Classify (ionic, covalent or polyatomic) and name the following compounds.

- CO<sub>2</sub> \_\_\_\_\_
- N<sub>2</sub>O \_\_\_\_\_
- NaClO<sub>2</sub> \_\_\_\_\_

Draw the Lewis Dot Diagrams for

- Carbon
- Sodium

Make balanced ionic compounds from:

- Li + Cl \_\_\_\_\_
- Be + NO<sub>3</sub><sup>-</sup> \_\_\_\_\_
- Na + Ar \_\_\_\_\_
- Ca + O \_\_\_\_\_

With dot diagrams draw the covalent bond of O<sub>2</sub>

With dot diagrams the covalent bond of CH<sub>4</sub>

Matching:

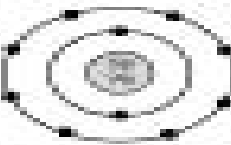
- Mixture
- Compound
- Element
- Matter
- Heterogeneous
- Homogeneous

- \_\_\_\_\_ A substance made up of two or more atoms that must be separated by chemical means.
- \_\_\_\_\_ Something made up of 2 or more different atoms that can be separated by physical means.
- \_\_\_\_\_ A substance that is pure and made up of only one type of atom.
- \_\_\_\_\_ Anything that has mass and takes up space.
- \_\_\_\_\_ A mixture that is the same throughout.
- \_\_\_\_\_ A mixture that is different throughout.

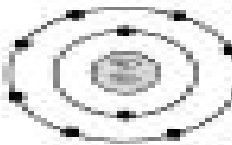
1. Valence electrons are the outermost electrons of an atom that are involved in chemical bonding. True/False

2. Lithium has \_\_\_\_\_ valence electrons, will \_\_\_\_\_ (lose or gain) \_\_\_\_\_ electrons and become \_\_\_\_\_ (positive or negative). This is why lithium's oxidation number is \_\_\_\_\_.

3. A metal and non-metal will form a \_\_\_\_\_ compound, two non-metals form a \_\_\_\_\_ compound.



Atom A



Atom B

Which one is the ion?

Give the ion notation for the ion.

Matching:

- Ion
- Element
- Isotope
- Oil

- \_\_\_\_\_ If you change the number of protons, you change the \_\_\_\_\_.
- \_\_\_\_\_ If you change the number of neutrons, you change the \_\_\_\_\_.
- \_\_\_\_\_ If you change the number of electrons, you change the \_\_\_\_\_.
- \_\_\_\_\_ Every 1,000 miles you should change a car's \_\_\_\_\_.

Matching:

- Unit of Measurement of Mass
- Meter
- Centimeter
- Liter
- Gram

- \_\_\_\_\_ Unit of length; equal to 0.1 m.
- \_\_\_\_\_ Mass is neither destroyed or created in chemical reactions.
- \_\_\_\_\_ Unit of area; about 1 dollar bill.
- \_\_\_\_\_ 1,000th of mass; width of pencil tip.
- \_\_\_\_\_ Unit of volume; just bigger than a quart.

Matching:

- Solid
- Liquid
- Gas
- Oil

- \_\_\_\_\_ Molecules that are tightly packed and retain their shape and size.
- \_\_\_\_\_ Molecules that bounce off of each other, can be compressed, and take the shape of their container.
- \_\_\_\_\_ Molecules that can move (slide) over each other and have a definite size (volume), but not shape and can not be compressed.

# Ipc Chemistry Fall Semester Review Answers

**Wolfgang Guggemos**



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