



# Manual Do Telefone Glt 2678

**R Sandford**



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web a carbon 12 12 c b this atom contains six protons and six neutrons there are six electrons in a neutral 12 c atom the net charge of such a neutral atom is zero and the mass number is 12 c the preceding answers are correct d the atom will be stable since c 12 is a stable isotope of carbon e the preceding answer is correct

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different elements will have the give the symbol and number of protons in one atom of atomic number

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13 the mass number is the number of protons and neutrons in the nucleus for this question you add the 4 protons and 5

neutrons to get an answer of 9 2 the lowest principal quantum number for an electron is 0 1 2 3

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web a both the electron geometry and the molecular structure are octahedral b both the electron geometry and the

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web a an oxygen atom with 8 protons and 8 neutrons b a potassium atom with 19 protons and 20 neutrons c a lithium atom

with 3 protons and 4 neutron give complete symbols of each atom including the atomic number and the mass number

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web exams prep master atomic structure incorporates three elements that wholly make an atom and they are named electron

proton and neutron in the structure of an atom the nucleus sets in the center and it contains the proton whereas electrons lie

on the periphery of the atom in an orbital way the first atomic model was given by j j thomson

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all matter the mass relationships between elements and compounds in chemical reactions ultimately relate back to the

characteristics of the atoms of which they are composed

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web atoms of the same element i e same atomic number  $Z$  that have different numbers of neutrons are called isotopes for example  $^{12}\text{C}$  of the carbon atoms on earth have 6 neutrons and 6 protons in their nuclei about 1% of the carbon atoms have 7

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web aug 19 2022 atoms with the same number of protons but different numbers of neutrons are called isotopes they share almost the same chemical properties but differ in mass and therefore in physical properties there are stable isotopes which do not emit radiation and there are unstable isotopes which do emit radiation

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web atoms of the same element that differ in their numbers of neutrons are called isotopes many isotopes occur naturally usually one or two isotopes of an element are the most stable and common different isotopes of an element generally have the same physical and chemical properties because they have the same numbers of protons and electrons

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web verified by toppr correct option is d atoms having same atomic number but different atomic mass are known as isotopes the difference in the mass is due to the difference in their number of neutrons with the same number of electrons and protons

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