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R Sanford



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web a carbon 12 12 c b this atom contains six protons and six neutrons there are six electrons in a neutral 12 c atom the net charge of such a neutral atom is zero and the mass number is 12 c the preceding answers are correct d the atom will be stable since c 12 is a stable isotope of carbon e the preceding answer is correct

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different elements will have the give the symbol and number of protons in one atom of atomic number

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13 the mass number is the number of protons and neutrons in the nucleus for this question you add the 4 protons and 5

neutrons to get an answer of 9 2 the lowest principal quantum number for an electron is 0 1 2 3

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web a both the electron geometry and the molecular structure are octahedral b both the electron geometry and the molecular structure are trigonal bipyramid c both the electron geometry and the molecular structure are linear d both the electron geometry and the molecular structure are trigonal planar

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web atoms of the same element i e same atomic number Z that have different numbers of neutrons are called isotopes for example $^{12}_6\text{C}$ of the carbon atoms on earth have 6 neutrons and 6 protons in their nuclei about 1 of the carbon atoms have 7

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