IA lab work on spectrophotometric determianation of K_c of the Iron(3)Thiocyanate system

Fe³⁺(aq) + SCN⁻(aq)
$$\rightleftharpoons$$
 FeSCN²⁺(aq)
$$K = \frac{[FeSCN^{2+}]}{[Fe^{3+}][SCN^{-}]}$$

Since I will not change the temperature in my experiment, the equalibrium constant will stay the same, and therefore I can measure the light absorbance and use the Lambert Beer-Law to find the initial concentartion of each species in the system and then calculate the K_c .

X(initial concentration of Iron3 and Thiocyanate)=A(absorbance of Iron(3)Thiocyanate solution)/E*I

Equipment:

-4 test tubes with 4 corks

-test tube rack

-10 ml pipitte with pump thing

Solutions:

-20 ml of Iron (3) solution

-20 ml of Thiocyanate solution

-20 ml of water H20

Samples	The amount of Thiocyanate solution SCN	The amount of water	Iron 3 solution
1	2 ml	8 ml	5 ml
2	4 ml	6 ml	5 ml
3	5 ml	5 ml	5 ml
4	9 ml	1 ml	5 ml

Kc Iron Thiocyanate Lab Report

Katrin Zwirglmaier

Kc Iron Thiocyanate Lab Report:

Annual Report - Brookhaven National Laboratory, 1960 Chemistry in the Laboratory Jo A. Beran, 1996 This updated and revised chemistry manual provides a well rounded understanding of concepts in the general chemistry laboratory Utilising visual aids experiments and equipment are explained and results and their pertinence discussed Proceedings of the Royal Irish Academy Royal Irish Academy, 1927 Includes also Minutes of the Proceedings and Report of the President and Council for the year beginning 1965 66 called Annual report **Proceedings** of the Royal Irish Academy .1927 Scientific and Technical Aerospace Reports ,1978 Zinc as a Galvanic **Anode** ,1956 Navy-industry Zinc Symposium on Cathodic Protection United States. Navy Department. Bureau of Navy-industry Zinc Symposium on Cathode Protection, Zinc as a Galvanic Anode, Development, Ships, 1956 **Application, and Specification, April 21-22, 1955** United States. Ships Bureau, 1957 **Reports on Public Health and Medical Subjects** Great Britain. Ministry of Health, 1968 Dictionary Catalog of the Research Libraries of the New York Public Library, 1911-1971 New York Public Library. Research Libraries, 1979 **Chemical Principles in the Laboratory** Emil J. Slowinski, 1996 Provides a series of experiments designed to teach students the available experimental methods the proper design of experiments and the interpretation of experimental results **Nutrition Abstracts and Reviews** ,1985 Geological Survey Research, Fiscal Year 1981 Geological Survey (U.S.),1984 Geological Survey Professional Paper Geological Survey (U.S.),1984 **Journal - American Water Works Association** American Water Works Association, 1931 U.S. Geological Survey Professional Paper ,1984 **Indian Science Abstracts** ,1990 Annual **Report** National Physical Laboratory (India),1966 Journal of Scientific & Industrial Research, 1963 Iournal of the American Chemical Society American Chemical Society, 1957 Proceedings of the Society are included in v 1 59 1879 1937

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