

Acid - Base Titration Lab

Knowledge		/16			
Introduction (Knowledge component)		/16			
-acids and bases are well defined		0	1	2	3
-strong and weak acid/base is clearly explained		0	1	2	3
-neutralization is well explained		0	1	2	3
-good definition of titration and explanation		0	1	2	3
-relevant real-world application of acid-base chemistry described		0	1	2	3
-purpose of the lab		0	1		
Communication		/12			
Title Page		/1			
-Includes appropriate and descriptive title for the lab		0	1		
Introduction (Communication component)		/4			
-written in full paragraphs (no headings/questions)		0	1	2	
-facts are correctly cited		0	1	2	
Results - Observation Tables		/5			
- Appropriate titles at the top		0	1		
- Units are in the headings		0	1		
- Well organized and neat (ex. numbers are aligned)		0	1		
- Contains all relevant data and observations (including averages)		0	1	2	
Overall clarity, neatness, formatting, spelling and grammar		/2			
Inquiry		/17			
Analysis		/4			
-Balanced chemical reaction is complete		0	1		
-Full calculations showing how to determine the concentration of HCl		0	1	2	3
Discussion		/9			
-difference between end point and equivalence point		0	1	2	
-NaOH's hygroscopic nature and its effect on the results		0	1	2	3
-Two experimental sources of error discussed and explained		0	1	2	3 4
Conclusion		/2			
-Final statement summarizing results of the lab		0	1		
-One thing that was learned from the lab		0	1		
References/Work Cited		/2			
-Correctly written in APA format					

Total : /45

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