# NACE CIP Level 2 Theory/Practical

Describe Passivation - correct answer Passivation is the formation of a protective oxide film on the surface reducing its chemical activity and its ability to corrode.

Describe the following factors and how they affect corrosion: - correct answer • Oxygen: Oxygen increases the rate of corrosion.

- Temperature: Corrosion usually accelerated with increasing temperature
- Chemical Salts: Increase the rate of corrosion by increasing the efficiency of the electrolyte.
- Humidity (or Wetness): The wetter the environment, the more corrosion is likely to occur.
- Pollutants and Acid Gases: Acid rain, chemical byproducts and chlorides all promote corrosion.

Two broad categories of corrosion can be described as: - correct answer General

Localized

Describe Galvanic Corrosion - correct answer Galvanic corrosion is an electrochemical action of two dissimilar metals in the presence of an electrolyte and an electron conductive path, which occurs when dissimilar metals come into contact.

Describe Cathodic Protection - correct answer Cathodic protection is the reduction or elimination of corrosion by making the structure to be protected a cathode by means of an impressed current or attachment to a galvanic anode.

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