

Name : \_\_\_\_\_

# Naming Acids and Bases

1) How can you tell by looking at a chemical formula that it is an acid?

\_\_\_\_\_

2) What is an oxyacid?

\_\_\_\_\_

Below are the formulas of several compounds. Circle the binary acids and put a square around the oxyacids

HCl    H<sub>2</sub>S    HMnO<sub>4</sub>    NaOH    HC<sub>2</sub>H<sub>3</sub>O<sub>2</sub>    HF    Sr(OH)<sub>2</sub>    H<sub>2</sub>SO<sub>4</sub>    NH<sub>3</sub>

3) Fill in the chart about naming binary acids.

Acid Formula	Positive Ion	Negative Ion	Acid Name
Example: HF	H <sup>+</sup>	F <sup>-</sup>	Hydrofluoric acid
HCl			
HBr			
HI			
H <sub>2</sub> S			
H <sub>2</sub> Se			

4) Fill in the chart about naming oxyacids.

Acid Formula	Positive Ion	Negative Ion	Acid Name
Example: H <sub>2</sub> SO <sub>4</sub>	H <sup>+</sup>	SO <sub>4</sub> <sup>2-</sup>	Sulfuric acid
HMnO <sub>4</sub>			
HCIO <sub>4</sub>			
HIO <sub>4</sub>			
HCIO			
C <sub>2</sub> H <sub>3</sub> O <sub>4</sub>			

# Naming Acids And Bases Evan P

**AW Rasmussen**



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