

Part F. Nomenclature of oxyacids – the -ic acids and the -ous acids

Table 7, Classification of types of compounds

Formula of acid	Anion	Name of acid
H_2SO_4	SO_4^{2-} , sulfate	sulfuric acid
HClO_4	ClO_4^- , perchlorate	perchloric acid
H_2SO_3	SO_3^{2-} , sulfite	sulfurous acid
HNO_3	NO_3^- , nitrate	nitric acid
HBrO	BrO^- , hypobromite	hypobromous acid
H_2CO_3	CO_3^{2-} , carbonate	carbonic acid
HNO_2	NO_2^- , nitrite	nitrous acid

Naming Oxyacids (using table 7 above)

1. What's the identity of the cation for each of the acids?
2. What's the trend in the name of the acids of the –ate oxyanions?
3. What's the trend in the name of the acids of the –ite oxyanions?
4. Write a general statement that describes how to predict the formula and name of an oxyacid given the formula and name of the oxyanion.

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