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$$\begin{array}{l} 1 \text{ N}_2 + 3 \text{ H}_2 \rightarrow 2 \text{ NH}_3 \\ 3 \text{ Fe} + 2 \text{ O}_2 \rightarrow \text{Fe}_3\text{O}_4 \\ 2 \text{ KClO}_3 \rightarrow 2 \text{ KCl} + 3 \text{ O}_2 \\ 2 \text{ NaCl} + \text{F}_2 \rightarrow 2 \text{ NaF} + \text{Cl}_2 \\ 4 \text{ H}_2 + \text{O}_2 \rightarrow 2 \text{ H}_2\text{O} \\ 5 \text{ Pb} + 2 \text{ OH}^- + 2 \text{ HCl} \rightarrow 2 \text{ H}_2\text{O} + 1 \text{ PbCl}_2 + 2 \text{ Al} + 3 \text{ K}_2\text{SO}_4 + 6 \text{ KBr} + 1 \text{ Al}_2\text{SO}_4 + 3 \text{ H}_2\text{O} + 2 \text{ CO}_2 + 4 \text{ CH}_4 \end{array}$$

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