

CBSE PAPER-2024 (SET-1)
CHEMISTRY

- The molar ionic conductivities of Ca^{2+} and Cl^- are 119.0 and $76.3 \text{ S cm}^2 \text{ mol}^{-1}$ respectively. The value of limiting molar conductivity of CaCl_2 will be : [1]
 (a) $195.3 \text{ S cm}^2 \text{ mol}^{-1}$ (b) $43.3 \text{ S cm}^2 \text{ mol}^{-1}$
 (c) $314.3 \text{ S cm}^2 \text{ mol}^{-1}$ (d) $271.6 \text{ S cm}^2 \text{ mol}^{-1}$
- Consider the following reaction : [1]

$$\begin{array}{c} \text{H} \\ \diagdown \\ \text{C} = \text{O} \\ \diagup \\ \text{H} \end{array} + \begin{array}{c} \text{H} \\ \diagdown \\ \text{C} = \text{O} \\ \diagup \\ \text{H} \end{array} + \text{Conc. KOH} \xrightarrow{\Delta} \text{A} + \text{B}$$

Identify A and B from the given options :

 (a) A – Methanol, B – Potassium formate (b) A – Ethanol, B – Potassium formate
 (c) A – Methanol, B – Ethanol (d) A – Methanol, B – Potassium acetate
- Which of the following acids represents Vitamin C ? [1]
 (a) Saccharic acid (b) Gluconic acid
 (c) Ascorbic acid (d) Benzoic acid
- Rosenmund reduction is used for the preparation of Aldehydes. The catalyst used in this reaction is : [1]
 (a) $\text{Pd} - \text{BaSO}_4$ (b) Anhydrous AlCl_3
 (c) Iron (III) oxide (d) HgSO_4
- Which alkyl halide from the given options will undergo $\text{S}_{\text{N}}1$ reaction faster ? [1]
 (a) $(\text{CH}_3)_3\text{C}-\text{Br}$ (b) $(\text{CH}_3)_2\text{CH}-\text{Br}$ (c) $\text{CH}_3-\text{CH}_2-\text{Br}$ (d) $(\text{CH}_3)_3\text{C}-\text{CH}_2-\text{Br}$
- From the elements of 3d series given below, which element shows the maximum number of oxidation states ? [1]
 (a) Scandium (b) Manganese (c) Chromium (d) Titanium

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1. Understanding the eBook Lgcse Adexel Chemistry Paper1 January 2014
 - The Rise of Digital Reading Lgcse Adexel Chemistry Paper1 January 2014
 - Advantages of eBooks Over Traditional Books
2. Identifying Lgcse Adexel Chemistry Paper1 January 2014
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Lgcse Adexel Chemistry Paper1 January 2014
 - User-Friendly Interface
4. Exploring eBook Recommendations from Lgcse Adexel Chemistry Paper1 January 2014
 - Personalized Recommendations
 - Lgcse Adexel Chemistry Paper1 January 2014 User Reviews and Ratings
 - Lgcse Adexel Chemistry Paper1 January 2014 and Bestseller Lists
5. Accessing Lgcse Adexel Chemistry Paper1 January 2014 Free and Paid eBooks
 - Lgcse Adexel Chemistry Paper1 January 2014 Public Domain eBooks
 - Lgcse Adexel Chemistry Paper1 January 2014 eBook Subscription Services
 - Lgcse Adexel Chemistry Paper1 January 2014 Budget-Friendly Options
6. Navigating Lgcse Adexel Chemistry Paper1 January 2014 eBook Formats

- ePub, PDF, MOBI, and More
 - Lgcse Adexel Chemistry Paper1 January 2014 Compatibility with Devices
 - Lgcse Adexel Chemistry Paper1 January 2014 Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Lgcse Adexel Chemistry Paper1 January 2014
 - Highlighting and Note-Taking Lgcse Adexel Chemistry Paper1 January 2014
 - Interactive Elements Lgcse Adexel Chemistry Paper1 January 2014
 8. Staying Engaged with Lgcse Adexel Chemistry Paper1 January 2014
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Lgcse Adexel Chemistry Paper1 January 2014
 9. Balancing eBooks and Physical Books Lgcse Adexel Chemistry Paper1 January 2014
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Lgcse Adexel Chemistry Paper1 January 2014
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Lgcse Adexel Chemistry Paper1 January 2014
 - Setting Reading Goals Lgcse Adexel Chemistry Paper1 January 2014
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Lgcse Adexel Chemistry Paper1 January 2014
 - Fact-Checking eBook Content of Lgcse Adexel Chemistry Paper1 January 2014
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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