

## Mole and Stoichiometry Mini-Lab

Calculations: Show **ALL** work (bridge/grid).

The **GIVEN** for calculations #4 and #5 should be measurements, not results from previous calculations.

4. How many moles of water are in the graduated cylinder?  
 UNKNOWN: mol  $\text{H}_2\text{O}$       GIVEN: g  $\text{H}_2\text{O}$  (data table)
5. How many molecules of water are in the graduated cylinder?  
 UNKNOWN: molecules  $\text{H}_2\text{O}$   
 GIVEN: g  $\text{H}_2\text{O}$  (data table)
6. For an experiment, 4.75 moles of water are needed.
  - a) How many grams of water are needed?
  - b) How many milliliters of water would this be?  
 UNKNOWN: g  $\text{H}_2\text{O}$       GIVEN: 4.75 mol  $\text{H}_2\text{O}$   
 UNKNOWN: mL  $\text{H}_2\text{O}$       GIVEN: 4.75 mol  $\text{H}_2\text{O}$

# Mini Lab 1 apply Stoichiometry

**Irene Cesa**



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**Recent Developments in Oxygen Minimum Zones Biogeochemistry** Sudheesh Valliyodan, Syed Wajih Ahmad Naqvi, Gireesh Kumar T. R., Annie Bourbonnais, Mark A. Altabet, Amal Jayakumar, 2023-12-18 **Stoichiometry Unit Project**  
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