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(i) Ethanoic acid,  $\text{CH}_3\text{COOH}$ , is a weak acid.

(ii) Complete the dot-and-cross diagram in Fig. 4.1 of a molecule of ethanoic acid.

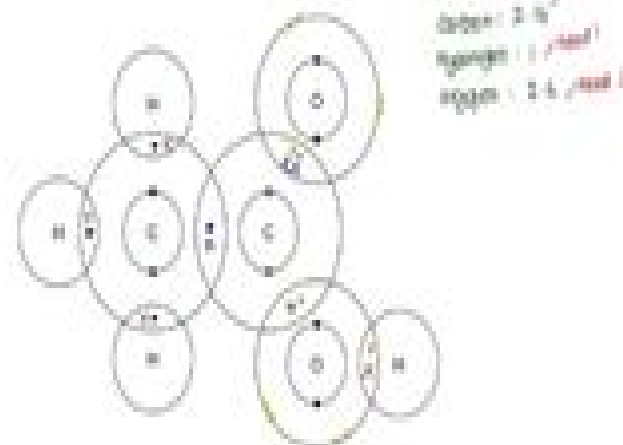


Fig. 4.1

(iii) Suggest the pH of dilute ethanoic acid.

3 < pH < 6

(iv) Complete the symbol equation to show the dissociation of ethanoic acid.



(v) Write the ionic equation for the reaction when an acid reacts with a soluble base.



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